664. Synthesis of 6-Halogenoflavones and Related Compounds. By C. T. CHANG, F. C. CHEN, T. S. CHEN, K. K. HSU, T. UENG, and (MISS) M. HUNG.

Preparation of several new 6-halogeno-flavones, -flavanones, and -flavonols, and corresponding chalcones is described. The nitration of o-hydroxyacetophenone and the Friedel-Crafts reaction of N-acetyl-p-anisidine, expected to yield the acetophenone required as the starting material for the synthesis of 6-iodo-compounds, have been investigated.

In continuation of previous studies on the halogenoflavones,¹ further compounds were synthesized several years ago.²⁻⁵ 6-Fluoro-,⁶ 6-chloro-,⁷ and 6-bromo-flavone⁸ were prepared in good yield in the usual way from 2-hydroxy-5-halogenoacetophenones which were obtained by Fries rearrangement of p-halogenophenyl acetates. However, later, in this Laboratory,⁴ attempts to prepare 2-hydroxy-5-iodoacetophenone by this method gave mainly 2,4-di-iodophenol.⁹ The iodoacetophenones have now been obtained by the two routes shown in the charts.



 ¹ Chen and Chang, J., 1958, 146.
 ² F. Chen, D.Sc. Dissertation, 1949, Tôhoku Univ., Sendai, Japan; Chen and Yang, J. Taiwan Pharm. Assoc., 1951, 3, 39.

³ Chen and Chang, J. Formosan Sci., 1954, 8, 74; Chen and Shu, J. Taiwan Pharm. Assoc., 1953, 5, 49.

- 4 T. S. Chen, B.Sc. Thesis, 1956, Nat. Taiwan Univ., Formosa.
- ⁶ Chen, Chang, and Chen, J. Formosan Sci., 1958, 12, 151; 1959, 13, 94.
 ⁶ Chen, Chang, Hsu, and Lin, J. Formosan Sci., 1954, 8, 23.
 ⁷ Chen and Hsu, J. Formosan Sci., 1953, 7, 51.
 ⁸ Chen, Lai, and Hsieh, J. Formosan Sci., 1953, 7, 57.
 ⁹ Chen Lai, and Hsieh, J. Formosan Sci., 1953, 7, 57.

- ⁹ Chang and Chen, J. Chinese Chem. Soc., Series II, 1960, 7, 69.

The Friedel-Crafts reaction of N-acetyl-p-anisidine with acetyl chloride in carbon disulphide at 25° afforded 83% of 2-methoxy-5- (I) and only 4% of 2-hydroxy-5-acetamidoacetophenone (II), but at 90° these proportions were reversed.

The 2-hydroxy-5-halogenoacetophenones condensed smoothly with benzaldehyde or p-anisaldehyde in presence of alcoholic alkali, giving chalcones, which were cyclized in phosphoric acid to 6-halogenoflavanones. These were easily dehydrogenated by N-bromosuccinimide to 6-halogenoflavones, which were also obtained by direct oxidation of the chalcones by selenium dioxide in pentyl alcohol.¹⁰ The 6-halogenoflavonols^{3b} were prepared in good yield by cold alkaline hydrogen peroxide.^{11,12} When this present work was complete, 5,13 we noted the synthesis of 6-iodoflavone and related compounds in other laboratories.14

EXPERIMENTAL

M. p.s were determined in capillary tubes and are uncorrected. Ethanol was used for crystallisation otherwise stated. Microanalyses were made by Mr. E. Aoyagi, Miike Branch, Mitsui Chemical Laboratory, Ohmuda, Japan.

2-Hydroxy-5-fluoro-, -chloro-, and -bromo-acetophenone.--p-Fluoro-, p-chloro-, and p-bromophenyl acetates were treated with anhydrous aluminium chloride, yielding colourless acetophenones as shown in Table 1.

2-Hydroxy-5-iodoacetophenone.-Method I. Nitration of o-hydroxyacetophenone by the procedure of Allan and Loudon ¹⁵ gave steam-volatile 2-hydroxy-5-nitroacetophenone, m. p.

Table 1.	Fries rearrangement of	ⁱ p-halogenopi	henyl	acetates to	2-hydr	·oxy-
	5-halogen	oacetophenon	es.			

<i>p</i> -Halogeno-	AlCl.		Time		Product
phenyl acetate	(g.)	Temp.	(hr.)	'Yield (%)	M. p. (lit.)
F 30 g.	45	150°	1	88	$57-58 \ (62\%, 56-56.5^{\circ})$
Cl 30 g.	45	120	1	70	$53-54$ ($52-54\%$, b 54 , c 55 , d 57
Br 30 g.	45	110	1	50	59-60 (57%, f 61-62)

^a Suter, Lawson, and Smith, J. Amer. Chem. Soc., 1939, **61**, 161. ^b Karrer, Yen, and Reichstein, Helv. Chim. Acta, 1930, **13**, 1315. ^c Claus, D.R.-P. 96,659. ^d Auwers and Wittig, Ber., 1924, **57**, 1275. ^e Nencki and Stoeber, *ibid.*, 1897, **30**, 1768. ^f Klarmann, Gates, Shternov, and Cox, J. Amer. Chem. Soc., 1933, 55, 4657. Feuerstein and Kostanecki, Ber., 1898, 31, 716, 2953.

2'-Hydroxy-		Yield	Fo	und (?	%)		Rec	uired	(%)
chalcone *	М.р.	(%)	С	н	Hal	Formula	С	н	Hal
5′-F	103—104°	83	$74 \cdot 15$	4 ·89	_	C ₁₅ H ₁₁ FO ₂	74 ·4	4 ·6	7 ·9
5′-Cl	111—112 ^h	90	—	—	$13 \cdot 8$	$C_{15}H_{11}ClO_{2}$	69.6	4.3	13.7
5′-Br	109·5—110 ⁱ	92	59.4	3∙8	$26 \cdot 4$	$C_{15}H_{11}BrO_{2}$	59.4	3.7	26.4
5′-I	107—108 ^j	86	51.2	3.3	35.5	$C_{15}H_{11}IO_{2}$	51.5	$3 \cdot 2$	36.2
5-F-4-MeO	125 - 126	74	77.15	4 ·9	—	C ₁₆ H ₁₅ FO ₃	70.6	4 ∙8	7.0
5'-Cl-4-MeO	111 - 112	71	$66 \cdot 2$	4 ·9	12.1	C ₁₆ H ₁₃ ClO ₃	66.6	4.5	12.3
5'-Br-4-MeO	115 - 116	90	57.55	4 ·1	$23 \cdot 8$	C ₁₆ H ₁₃ BrO ₃	57.7	3.9	24.0
5'-I-4-MeO	130	70	50.45	3.75	33 ·6	C ₁₆ H ₁₃ IO ₃	50.6	3.5	33.4

* All yellow needles. ^h Ref. b of Table I gives m. p. 108°; Kinckell (Ber., 1911, **44**, 3654) gives m. p. 109–110°. ⁱ Kostanecki and Ludwig (Ber., 1898, **31**, 2951) give m. p. 107–108°. ^j Lit.,¹⁴ m. p. 138°.

101-102° (29%) (lit.,¹⁵ 98-99°, 19%), and involatile 2-hydroxy-3-nitroacetophenone, m. p. $84-84\cdot5^{\circ}$ (1.9%) (lit., ¹⁵ 82-83°, 1.9%), which were reduced with stannous chloride and hydrochloric acid, yielding 5-amino-2-, m. p. 112-113° (Found: C, 63.0; H, 6.4; N, 9.1. C₈H₉O₂N requires C, 63.6; H, 6.0; N, 9.3%), and 3-amino-2-hydroxyacetophenone, m. p. 93-94° (Found: C,

¹⁰ Mahal, Rai, and Venkataraman, J., 1935, 866; 1936, 569.
 ¹¹ Algar and Flynn, Proc. Roy. Irish Acad., 1934, 42, B, 1.

12 Oyamada, J. Chem. Soc. Japan, 1934, 55, 1256.

¹³ Chen, Chang, Hung, Lin, and Choong, Proc. Chem. Soc., 1959, 232.
 ¹⁴ Mulchandani and Shah, Chem. Ber., 1960, 98, 1913.

¹⁵ Allan and Loudon, J., 1949, 821.

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63·2; H, 6·3; N, 9·2%), all yellow needles. Diazotisation, etc., then yielded 2-hydroxy-5-(75%), m. p. 90—91° (lit., ¹⁴ 90°) (Found: C, 37·1; H, 2·9; I, 50·1. Calc. for $C_8H_7IO_2$: C, 36·5; H, 2·7; I, 48·4%), and 2-hydroxy-3-iodoacetophenone (50%), m. p. 58—59° (Found: C, 36·22; H, 2·91; I, 48·2%), both as colourless needles.

Method II. (a) Acetylation at 80–90°. To a solution of N-acetyl-p-anisidine (20 g.) and acetyl chloride (25 c.c.) in dry carbon disulphide (50 c.c.) was added anhydrous aluminium chloride (55 g.) in portions, with stirring, and the mixture was kept at 80–90° for 1.5 hr., while

TABLE 3.	6-Halogenoflavanones	and the $4'$ -methox	y-compounds
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		Yield	Fo	ound (%)		Req	uired	(%)
Flavanone *	М.р.	(%)	С	н	Hal	Formula	С	н	Hal
6-Fl	77—78°	70	74 ·1	4 ·9	_	C ₁₅ H ₁₁ FO ₂	74 ·4	4 ·6	7.9
6-Cl- †	96—97	50	_		13.9	$C_{15}H_{11}ClO_{9}$	69.6	4 ·3	13.7
6-Br	118	60	—	—	$26 \cdot 4$	$C_{15}H_{11}BrO_{2}$	59.4	3.7	26.4
6-I	137 *	91	51.5	3.4	34.9	C ₁₅ H ₁₁ IO,	51.5	$3 \cdot 2$	36.2
6-F-4'-MeO	88—89	50	70.2	5.0	—	C ₁₆ H ₁₅ FO ₅	70.6	4 ·8	7.0
6-Cl-4'-MeO	$106 - 106 \cdot 5$	45	66·4	4 ·7	12.15	C ₁₆ H ₁₃ ClO ₃	66.6	4.5	12.3
6-Br-4'-MeO	111.5 - 112.5	40	57.3	$4 \cdot 2$	$23 \cdot 8$	$C_{16}H_{13}BrO_{3}$	57.7	3.9	24.0
6-I-4'-MeO	112 '	58	50.4	3.9	$33 \cdot 2$	$C_{16}H_{13}IO_3$	50.6	$3 \cdot 5$	33 ·4

[†] Prisms; previously reported by Karrer, Yen, and Reichstein (*Helv. Chim. Acta*, 1930, **13**, 1315) who give no m. p. ^k Lit., ¹⁴ m. p. 136°. ^l Lit., ¹⁴ m. p. 101°.

TABLE 4. 6-Halogenoflavones	and the 4'-methoxy-compounds
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		Yield ^q	Fo	und (9	%)		Req	uired	(%)
Flavone *	М.р.	(%)	С	н	Hal	Formula	С	н	Hal
6-F	128—129°	53	74 ·8	4 ·0		C ₁₅ H ₉ FO ₂	75.0	3 ∙8	7.9
6-Cl	182—183 ^m	61	—	—	13.9	C ₁₅ H ₆ ClO ₉	70.2	3.5	13 ·8
6-Br	189—190 ⁿ	42	—		26.5	C ₁₅ H ₆ BrO,	59.8	3 ∙0	26.5
6-I	190-191 °	74 P	51.7	$2 \cdot 9$	36.4	C ₁₅ H ₆ IO,	51.8	2.6	36 ·5
6-F-4'-MeO	159 - 160	50	70.6	4 ·8	—	C ₁₆ H ₁₁ FO ₃	$71 \cdot 1$	4 ·1	7.0
6-Cl-4'-MeO	180-181	77	66.9	4 ·1	12.3	C ₁₆ H ₁₁ ClO ₃	67.0	3.9	12.4
6-Br-4'-MeO	192—193	76	57.9	3.7	$23 \cdot 9$	C ₁₆ H ₁ ,BrO ₅	58.0	3.4	$24 \cdot 1$
6-I-4'-MeO	183	71	50.7	$3 \cdot 2$	33.3	C ₁₆ H ₁₁ IO ₃	50.8	$2 \cdot 9$	33 .6

* Colourless needles. ^m Wittig, Bangert, and Richter (Annalen, 1925, **446**, 155) give m. p. 181°; Ruheman (Ber., 1921, **54**, 912) gives m. p. 183—184°. ⁿ Ref. *i* of Table 2 gives m. p. 189—190°; Bhagwat and Wheeler (J., 1939, 96) give m. p. 192—193°. ^o Lit., ¹⁴ m. p. 184°. ^p Lit., ¹⁴ m. p. 167°. ^e All obtained by the selenium dioxide method; the last compound was also prepared in 55% yield by the N-bromosuccinimide method.

carbon disulphide was distilled off. After decomposition with ice-water, the product was collected and extracted with 5% aqueous sodium hydroxide. After acidification of the alkaline solution with hydrochloric acid, the precipitated 5-acetamido-2-hydroxyacetophenone (II) was collected and crystallised as pale yellowish crystals (19.2 g., 82%), m. p. $167-168^{\circ}$ (lit., ¹⁶ 165°).

The alkali-insoluble product was recrystallised, giving 5-acetamido-2-methoxyacetophenone (0.7 g., 3%), m. p. $189-190^{\circ}$ (lit., ¹⁶ $186-188^{\circ}$).

(b) Acetylation at 25°. N-Acetyl-p-anisidine (20 g.) was treated as above but at $20-25^{\circ}$ for 12 hr., then worked up as above, yielding 5-acetamido-2-hydroxyacetophenone (I) (1 g., 4%) and 5-acetamido-2-methoxyacetophenone (II) (20.5 g., 83%).

5-Acetamido-2-methoxyacetophenone (II, 5 g.), anhydrous aluminium chloride (5 g.), and chlorobenzene (25 c.c.) were boiled for 1 hr., then decomposed with ice-water. The chlorobenzene was steam-distilled off, and the precipitate was collected and extracted with 10% aqueous sodium hydroxide. The precipitate formed from the alkaline solution on acidification was collected and crystallised, giving 5-acetamido-2-hydroxyacetophenone (II) (4.3 g., 90%).

5-Amino-2-hydroxyacetophenone.—5-Acetamido-2-hydroxyacetophenone (II) (20 g.) was boiled with 15% hydrochloric acid (25 c.c.) for 40 min., then neutralized with aqueous ammonia and crystallized from 10% ethanol, affording 5-amino-2-hydroxyacetophenone (13·2 g., 84%), yellow needles, m. p. 112—113° (Found: C, 63·0; H, 6·4; N, 9·1. $C_8H_9O_2N$ requires C, 63·6; H, 6·0; N, 9·3%); its hydrochloride formed colourless plates, m. p. 230—240° (decomp.) (lit.,¹⁴ 118°).

¹⁶ Mathieson and Newberry, *J.*, 1949, 1135.

2-Hydroxy-5-iodoacetophenone.—5-Amino-2-hydroxyacetophenone (5 g.) and ice (20 g.) in 18N-sulphuric acid (18 c.c.) were treated with sodium nitrite (2·4 g.) in water (4 c.c.). When diazotization was complete, more concentrated sulphuric acid (3 c.c.), cupper powder (0·1 g.), and potassium iodide (6·3 g.) in water (10 c.c.) were added. Then the temperature was slowly

TABLE 5.	6-Halogenofla	wonols and	the 4'-met	hoxy-compounds.*	4
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	Found (%)						uired	Acetate,	
Flavonol †	М.р.	С	H	Hal	Formula	С	н	Hal	m. p.
6-F	$163 - 165^{\circ}$	70.1	3.75	—	C ₁₅ H ₉ FO ₃	70·3	3.5	7.4	
6-Cl	162—163 ^r	66 ·0	3.55	$12 \cdot 8$	C ₁₅ H ₆ ClO ₃	66·1	3.3	13·0	
6-Br	183—184	56.7	3.1	25.0	C ₁₅ H ₉ BrO ₃	56.8	$2 \cdot 9$	25.2	143—144°
6-I	191—192 *	49·3	$2 \cdot 9$	34.5	C ₁₅ H ₆ IO ₆	49.5	$2 \cdot 5$	34 ·9	175 - 176
6-F-4'-MeO	207 - 208	66.9	4.25		C, H, FO	67.1	3.9	6.6	
6-Cl-4'-MeO	205 - 206	63·3	3.9	11.4	C, H, ClO	63.5	3.7	11.7	160-161
6-Br-4'-MeO	193—194	$55 \cdot 2$	3 ∙5	22.7	C, H, BrO	$55 \cdot 4$	$3 \cdot 2$	$23 \cdot 0$	
6-I-4'-MeO	179—180 *	4 8·7	3 ∙0	$32 \cdot 0$	$C_{16}H_{11}IO_{4}$	48 ·8	2.8	$32 \cdot 2$	187—188

* Preliminary reports; Chen and Shu, J. Taiwan Pharm. Assoc., 1953, 5, 49. † Yellow needles. * Minton and Stephen (J., 1922, 121, 1598) give m. p. 163°. * Lit.,¹⁴ m. p. 194°. * Lit.,¹⁴ m. p. 186°.

raised and kept at 65° until no more nitrogen was evolved. Steam-distillation and recrystallization yielded 2-hydroxy-5-iodoacetophenone (7 g., 80%) as colourless needles, m. p. $91-92^{\circ}$, identical with the product obtained by method I.

The 6-halogeno-flavones, -flavonoes, -flavonols, and corresponding chalcones were prepared as described in a previous paper,¹ and are listed in Tables 2-5.

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